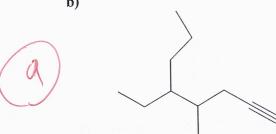
1. Write the IUPAC name of the following compounds in the box provided

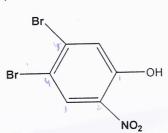
a)

$$CI$$
 CH_3 CH_2 CH_3 CH_2 CH_2 CH_3

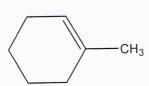
b)

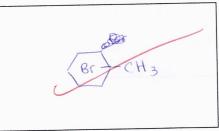


c)



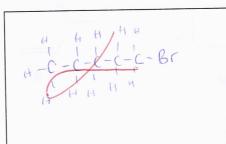
- 2. Draw the structure of:
 - a) The compound of molecular formula C₇H₁₃Br that will give the following alkene as the exclusive product of E2 elimination





b) The compound of molecular formula C₅H₁₁Br that is **not capable** of undergoing E₂ elimination

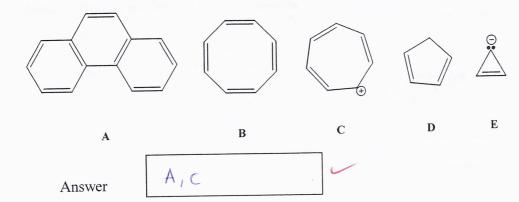




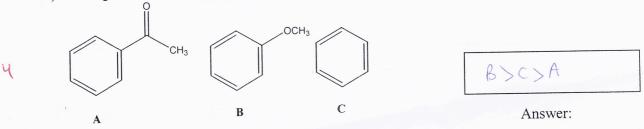
3. Answer the following questions in the box provided

6

a) Which of the following compounds or ions are <u>aromatic</u> species?



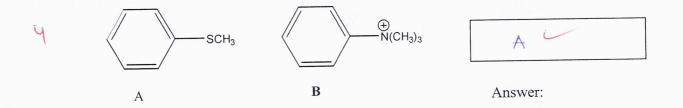
b) Arrange in order of decreasing reactivity with SO₃ (most reactive first)



c) Draw the structure of the <u>monochlorination</u> product of the following compound with Cl₂ and FeCl₃

$$H_3C$$
 CF_3
 CF_3

d) Which compound will add a nitro group to the <u>ortho</u> and <u>para</u> positions?



4. Give the structure of the product of each of the following reactions. If more than one product is obtained indicate which one is the major product.

Include stereochemistry where applicable

d) Br
$$\frac{1.\text{NaNH}_{2}, \text{NH}_{3}}{2.\text{ H}_{2}\text{O}}$$

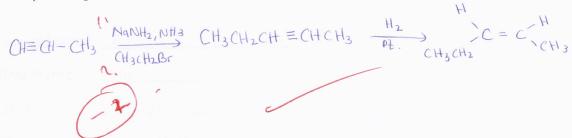
e)
$$CH_3CH_2C$$
 $CCH_2CH_3 + H_2O$ $HgSO_4$

(15

$$\begin{array}{c} \text{CH}_3 \\ \text{f)} \\ \text{CI} \\ \text{NaOCH}_2\text{CH}_3 \\ \text{CH}_3\text{CH}_2\text{OH} \\ \text{H}_2\text{O},\text{heat} \\ \text{II} \\ \text{FeBr}_3 \\ \text{II} \\ \text{II} \\ \text{CH}_3 \\ \text{CH$$



- **5.** Outline an efficient synthesis of each of the following compounds from the indicated starting material and any necessary organic or inorganic reagents
- a) cis-2-pentene from propyne



- (6)
- b) *m*-Bromoacetophenone from benzene

b) m-Bromoacetophenone from behizelie
$$+ CH_3CCL + Br_2$$

$$+ Br_2$$

$$+ Br_2$$

$$+ Br_3$$

Bonus questions (5pts)

a) Draw the structure of the <u>most stable</u> resonance contributing structure of the following intermediate:



b) How many benzylic hydrogens are present in the following hydrocarbon

